

From the diluted regime to the concentrated regime :

# the case of spherical systems (colloids, spherical micelles, globular proteins..)

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What is the structure at different scales ?



F. Cousin et al, Langmuir, 2010, 26(10), 7078–7085.









$$\sum(q)$$

$$\sum(q)$$

$$(\rho_{obj} - \rho_{media})^2 v_{obj}^2 = n \sum_{i,j}^N e^{i\vec{q}\cdot(\vec{r}_i^\alpha - \vec{r}_j^\alpha)} + n^2 \sum_{i,j,\alpha\neq\beta}^N e^{i\vec{q}\cdot(\vec{r}_i^\alpha - \vec{r}_j^\beta)}$$

$$P(q) = \frac{1}{N^2} \sum_{i^\alpha}^N \sum_{j^\alpha}^N e^{i\vec{q}\cdot(\vec{r}_i^\alpha - \vec{r}_j^\alpha)}$$

$$Q(q) = \frac{1}{N^2} \sum_{i^\alpha}^N \sum_{j^\beta}^N e^{i\vec{q}\cdot(\vec{r}_i^\alpha - \vec{r}_j^\beta)}$$

*Intra correlations* : **Form factor P(q)** of objects (Mass, Size, shape, gyration radius...)

Inter correlations : Structure factor Q(q) (ou S(q)) of objects (Interactions, 2<sup>nd</sup> Virial coefficient..)

Cf lecture Pierre Roblin



#### For rigid centrosymmetrical objects :





#### For rigid centrosymmetrical objects :



For no rigid centrosymmetrical objects : polymers, hydrogel, etc..



Cf lecture François Boué





- Measurement in a system without interactions
- Dilution of the system (extrapolation to zero concentration)
- Kill the interactions, by adding salt for instance (more risky!)



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#### Link between S(q) and correlation function g(r)



**g(r)** : Average of the normalized density of objects in a shell [r, r+dr] from the center of a particle

#### Link between S(q) and correlation function g(r)



(courtesy of a lecture by J.S. Pedersen)

$$S(q) = 1 + n 4\pi \int (g(r) - 1) \frac{\sin(qr)}{qr} r^2 dr$$

Progressive addition of salt on an electrostatically stabilized colloidal suspension





0.0025 M



Progressive addition of salt on an electrostatically stabilized colloidal suspension





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> Informations at low q ?





**Density fluctuations** at large scale  $(q \rightarrow 0)$ 

$$\left[\frac{\partial\Pi}{k_{B}T\partial\rho}\right]_{T} = \frac{1}{S(0)}$$





$$\left[\frac{\partial\Pi}{k_B T \partial\rho}\right]_T = \frac{1}{S(0)} = 1 + \rho A_2 + \rho^2 A_3 + \rho^3 A_4 + \dots$$





One can show that :

$$I(q) = \Phi \Delta \rho^2 V_{part} P(q).S(q)$$

**Determination of A<sub>2</sub>** : several measurements at concentrations at low q

At low q: 
$$P_{Guinier}(q) = 1 - \frac{q^2}{3} \left[ \frac{1}{2N^2} \sum_{i}^{N} \sum_{j}^{N} r_{ij}^2 \right] = 1 - \frac{q^2 R_g^2}{3}$$

$$\left[\frac{\partial\Pi}{k_B T \partial\rho}\right]_T = \frac{1}{S(0)} = 1 + \rho A_2 + \rho^2 A_3 + \rho^3 A_4 + \dots$$



One can show that : 
$$A_2 = \frac{1}{2} \int_0^\infty \left(1 - e^{-\frac{V(r)}{kT}}\right) 4\pi r^2 dr$$



How to go further in descriptions of interactions ?



Simulations, integral equations



See, for example

Chapter from Luc Belloni (in french)



















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 Different structure for liquid and solid glassy samples

#### Silica spheres in aqueous suspensions



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- Different structure for liquid and solid glassy samples
- Dilution law
- $\succ$  Link between S(0) and  $\Pi$

#### **Interacting micelles of surfactants**

Aqueous mixture of stearic acid and 12-hydroxystearic acid at fixed total concentration of surfactants





Be careful : aggregation number (and form factor) may change with physico-chemical conditions

#### Interacting micelles of surfactants

Aqueous mixture of stearic acid and 12-hydroxystearic acid at fixed total concentration of surfactants



- Be careful : aggregation number (and form factor) may change with physico-chemical conditions
- Fitting of structure factor by Hayter Penfold is usually successful

#### Correlation length $\boldsymbol{\xi}$ in attractive systems

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(optical microscopy)

Approach of critial point by temperature at fixed salinity

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Approach of critial point by temperature at fixed salinity

#### $\rightarrow$ Divergence of fluctuations

F. Cousin et al, J. Phys. Chem B, 2001, 115(13), 6051-6057

#### Correlation length $\boldsymbol{\xi}$ in attractive systems

Progressive addition of salt on an electrostatically stabilized colloidal suspension



$$g(r) \propto \frac{e^{-r/\xi}}{r} \implies S(q) \propto \frac{1}{q^2 + \xi^{-2}}$$

F. Cousin et al, J. Phys. Chem B, 2001, 115(13), 6051-6057





Fractal final structure (D<sub>f</sub> = 2.1)

**Diffusion Limited Cluster Aggregation** 

**Reaction Limited Cluster Aggregation** 

2 limit cases of aggregation processes :

(without strong attractions)





**Diffusion Limited Cluster Aggregation** 



Fractal final structure  $(D_f = 2.1)$ 

#### **Reaction Limited Cluster Aggregation**







#### **Dense aggregates**



Desalting (+ water)



S. Mehan et al, Soft Matter, 2021, 17, 8496-8505

#### **Dense aggregates**



Fit by Percus Yevick structure factor
 Determination of compactness of aggregates



 $Q(nm^{-1})$ 

(a)

S. Mehan et al, Soft Matter, 2021, 17, 8496-8505



Clusters of gold assembled by ligands



> Aggregation number  $N_{agg}$  from  $I(q_0)$ :  $N_{agg} = I_{cluster} (0)/I_{individual object} (0)$ 



#### **Bonus : anisotropic systems based on spheres**



## $S(q)_{//}$ or $S(q)_{\perp}$ allows to describe qualitatively the structure

A.S. Robbes et al, Macromolecules, 2022, 55(15), 6876-6889.

## 24<sup>th</sup>-25<sup>th</sup> January 2024 **Paul Scherrer Institute PSI** Villigen, Switzerland

Liquids, colloids, surfactants, polymers, foams, gels, granulars, liquid crystals, emulsions, capsules, proteins, biological materials, food, pharma, cosmetics, and more!

Abstract submission 10<sup>th</sup> Nov/23

STANS FOR SOFT MAT

Registration 1<sup>st</sup> Dec/23

#### Invited guests

Prof. Andrea Scotti, Malmö University, Malmö Dr Jean-Paul Chapel, CRPP, Bordeaux Dr. Isabelle Capron, INRAE, Nantes Prof. Harm-Anton Klok, EPFL, Lausanne Prof. Peter Fischer, ETH, Zurich Dr. Pierre Bauduin, ICSM, Marcoule

> Société Francais









A.S. Robbes et al, Macromolecules, 2012, 45, 9220–9231



Photons and neutrons do not see the same things...

![](_page_52_Figure_1.jpeg)

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Robbes et al, Soft Matter, 2012, 8, 3407-3418 A.S. Robbes et al, Macromolecules, 2012, 45, 9220-9231